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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/816,603	03/23/2001	Mark Lynn Jenson	1327.009US1	6175
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Schwegman, Lundberg,			EXAMINER	
Woessner & Kluth, P.A. P.O. Box 2938			ALEJANDRO, RAYMOND	
Minneapolis, MN 55402				
William Capolis, Will 33 102			ART UNIT	PAPER NUMBER
			1745	1,
			DATE MAILED: 01/22/2003	K

Please find below and/or attached an Office communication concerning this application or proceeding.

		A 5-		
	Application No.	applicant(s)		
	09/816,603	JENSON, MARK LYNN		
Office Action Summary	Examiner	Art Unit		
	Raymond Alejandro	1745		
The MAILING DATE of this communication apperiod for Reply	ppears on the cover sheet with	the correspondence address		
A SHORTENED STATUTORY PERIOD FOR REPORTED MAILING DATE OF THIS COMMUNICATION - Extensions of time may be available under the provisions of 37 CFR 1 after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a recommunication of the period for reply is specified above, the maximum statutory period Failure to reply within the set or extended period for reply will, by statuent of the provided patient than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b). Status	. 136(a). In no event, however, may a replicible. 1919 within the statutory minimum of thirty (3 d will apply and will expire SIX (6) MONTH 1ste, cause the application to become ABAN	y be timely filed 30) days will be considered timely. S from the mailing date of this communication. IDONED (35 U.S.C. § 133).		
1) Responsive to communication(s) filed on 20	December 2002 .			
	This action is non-final.	•		
3)☐ Since this application is in condition for allow		rs, prosecution as to the merits is		
closed in accordance with the practice unde	er <i>Ex parte Quayle</i> , 1935 C.D.	11, 453 O.G. 213.		
4)⊠ Claim(s) <u>11-41</u> is/are pending in the applicat	tion.			
4a) Of the above claim(s) 13,16,17,25-30,34-	38 and 41 is/are withdrawn fro	om consideration.		
5) Claim(s) is/are allowed.				
6)⊠ Claim(s) <u>11,12,14,15,18-24,31-33,39 and 40</u>	is/are rejected.	·		
7) Claim(s) is/are objected to.				
8) Claim(s) are subject to restriction and/	or election requirement.			
Application Papers				
9) The specification is objected to by the Examin		_		
10) The drawing(s) filed on is/are: a) acc	· · · · · · · · · · · · · · · · · · ·			
Applicant may not request that any objection to t 11) The proposed drawing correction filed on				
If approved, corrected drawings are required in r		approved by the Examiner.		
12) The oath or declaration is objected to by the E	• •			
Priority under 35 U.S.C. §§ 119 and 120				
13) Acknowledgment is made of a claim for foreign	an priority under 35 U.S.C. § 1	119(a)-(d) or (f)		
a) ☐ All b) ☐ Some * c) ☐ None of:	5			
1. Certified copies of the priority documer	nts have been received.			
2. Certified copies of the priority documents have been received in Application No				
 Copies of the certified copies of the pri application from the International B 	ority documents have been re Bureau (PCT Rule 17.2(a)).	ceived in this National Stage		
* See the attached detailed Office action for a list	·			
14) Acknowledgment is made of a claim for domes				
a) The translation of the foreign language parts) Acknowledgment is made of a claim for domes	' '			
Attachment(s)				
 Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Review (PTO-948) Information Disclosure Statement(s) (PTO-1449) Paper No(s) 	5) Notice of Info	mmary (PTO-413) Paper No(s) primal Patent Application (PTO-152)		

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DETAILED ACTION

Election/Restrictions

- 1. Applicant's election without traverse of Group 2 (claims 11-31) in Paper No. 3 is acknowledged.
- 2. It is also noted that applicant did not elect a single claimed species as identified in the Restriction Requirement, <u>paper # 2</u>. Accordingly, in order to expedite the prosecution of the instant application, a telephone call was made to Charles A. Lemaire on 01/13/03 to request an oral election to the foregoing restriction requirement as applicable to <u>the Election of Species</u>. It resulted that applicant further elected <u>Species 1- a photovoltaic cell on the battery</u>. Thus, an action as to this particular merit follows.

Claim Rejections - 35 USC § 112

- 3. The following is a quotation of the second paragraph of 35 U.S.C. 112:
 The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.
- 4. Claims 11-12, 14-15, 18-20, 31 and 33 is rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.
- 5. The language "ion-assist energy" in claims 11 and 33 is of uncertain meaning, thereby rendering the claim vague. Further, the language "ion-assist energy" is not defined by the claim, the specification does not provide a standard for ascertaining the requisite degree, and one of ordinary skill in the art would not be reasonably apprised of the scope of the invention. It is uncertain as to what is particularly intended to recite by the term "ion-assist".

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6. Claim 33 recites the limitations "a substrate having a major surface area" in line 2, and "the substrate" in line 3. There is insufficient antecedent basis for this limitation in the claim.

Claim 11 contains an earlier recitation of both limitations, therefore, it is unclear whether the limitations of claim 33 refer to a second substrate therein or to the same substrate as in claim 1.

Claim Rejections - 35 USC § 103

- 7. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 8. Claims 11-12, 14-15, 18-24, 31, 33 and 39-40 are rejected under 35 U.S.C. 103(a) as being unpatentable over Ovshinsky et al 5411592 in view of Bates et al 5338625.

The instant application is directed to system for making a thin-film device wherein the claimed inventive concept comprises the specific deposition means (station) that deposits the layers. Other limitations include: the photovoltaic cell on the battery; the motion device, the rolled flexible material; the rigid material; the polymer; the layer energizing; the continuous plastic sheet and wafers; the energy conversion device; the specific layer material; and the second substrate means.

With respect to claims 11, 21, 22 and 31:

Ovshinsky et al disclose an apparatus for deposition of thin-film solid state batteries (title) comprising a multi-chambered deposition apparatus for depositing battery materials onto substrate material (abstract/col 6, lines 25-43). The apparatus includes at least three distinct

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evacuable deposition chambers, interconnected in series; the first deposition chamber is adapted to deposit a layer of battery electrode material onto the substrate (abstract/col 6, lines 25-43). The second deposition chamber is adapted to deposit a layer of electrolyte material onto the layer of the battery electrode material deposited in the first chamber. The third deposition chamber is adapted to deposit another layer of battery electrode material onto the electrolyte layer (abstract/col 6, lines 25-43). Initially, the substrate passes to the first deposition chamber then it is transported to the second chamber, next the substrate is passed through another gas gate into the third deposition chamber (col 11, lines 58 to col 12, line 7). *Thus, the process is continuous*.

Each electrochemical cell includes a thin-film negative electrode layer, a thin-film positive electrode layer and a thin-film electrolyte layer (col 9, lines 25-28). The chambers are specifically adapted to deposit battery materials onto the substrate (col 11, lines 50-58). The energy conversion device is the battery itself which is being deposited over the substrate in the form of different layers.

The deposition chambers are preferably adapted to deposit materials by at least one method selected from the group consisting of chemical vapor deposition, microwave plasma enhanced chemical vapor deposition, sputtering, laser ablation among them (col 7, lines 65 to col 8, line 3). It is noted that sputtering and laser ablation are ion-assist energy deposition techniques.

As to claim 12:

It is disclosed that a product variation is the deposition of the thin-film batteries onto substrates on the opposite side of thin-film silicon solar cells (photovoltaic cells) to integrate the collection and storage of solar energy (col 11, lines 39-43).

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As to claim 14:

It is disclosed that the chambers are physically interconnected in series (col 6, lines 29-31) and the deposition chambers are interconnected by gas gates such that the substrate material is allowed to proceed from one deposition chamber to the next, while maintaining gaseous segregation between the chambers (col 6, lines 40-44).

In reference to claim 15:

It is taught that a third embodiment comprises an evacuable payout chamber which is physically connected in series to the first deposition chamber; the payout chamber holds a roll of substrate material (flexible material as it has been rolled) which is unrolled and passed to the first deposition chamber (col 6, line 65 to col 7, line 2).

On the matter of claims 18-19, 23 and 39:

Ovshinsky et al disclose that the substrate may be formed from an electrically conductive metal (rigid material) or from an electrically insulating polymer (col 9, lines 3-6). Thus, the rolled substrate material is understood to be a continuous plastic sheet. The use of an elongated web of substrate material is disclosed (col 13, lines 13-17).

With respect to claims 24 and 40:

It is disclosed that a second embodiment comprises a deposition apparatus for depositing single or multi-celled batteries upon precut substrates (wafers), that is a substrate which is of relatively limited length and width dimensions when compared to rolls of substrate web which can be as long as 2000 ft or more (col 12, lines 35-46).

As far as claim 32 (see also rejection below):

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Ovshinsky et al disclose that for lithium ion system, the positive electrode layer can be formed from a material such as metal oxides and lithiated metal oxides compound such as LiCoO₄ (col 10, lines 30-34).

Regarding claim 33:

It is disclosed that a second embodiment comprises an apparatus including an substrate insertion chamber which is physically interconnected in series to the first deposition chamber; the insertion chamber is adapted to hold one or more individual substrates and pass them to the first deposition chamber (col 6, lines 44-64).

Ovshinsky et al disclose an apparatus for deposition of thin-film batteries according to the foregoing. However, Ovshinsky et al do not expressly disclose the deposition station supplying the amount of ion-assist energy to aid in crystalline layer formation while controlling stoichiometry of the crystalline layer and the specific ion energy.

As to claims 11, 20, 21, 22:

Bates et al disclose a thin-film battery and method for making same wherein it is disclosed that the performance of the lithium battery is very dependent on formation of the cathode. Consideration of the microstructure of the cathode is equally important as consideration of the composition; typical is the use of cathodes having a characteristic crystalline microstructure which is dependent on the substrate temperature, extent of the erosion of the target material due to prior sputtering and the pressure and composition of the process gas during deposition (col 5, lines 25-34). At certain substrate temperatures, the battery cathodes consist of crystalline pellets while films deposited onto substrates consist of clusters of crystalline fibrous bundles (col 5, lines 35-39).

In view of these disclosures, it would have been obvious to one skilled in the art at the time the invention was made to use specific deposition station supplying the specific amount of ion-assist energy to aid in crystalline layer formation while controlling stoichiometry of the crystalline layer in the system of Ovshinsky et al as Bates et al disclose that that the performance of the lithium battery is very dependent on formation of the cathode wherein consideration of the microstructure of the cathode is equally important as consideration of the composition. It is typical to use cathodes having a characteristic crystalline microstructure which is dependent on the process substrate temperature, extent of the erosion of the target material due to prior sputtering and the pressure and composition of the process gas during deposition per se. Thus, those of ordinary skilled in the art can appreciate that by controlling and applying specific energy magnitudes onto the substrate material a crystalline substrate material formation is obtained, and thus, the performance of the battery is enhanced, and a cell of much higher capacity is also obtained. Further, since Bates et al also disclose that when deposited from an eroded target, the cathode films where characterized by a high density of micron-sized fibrous cluster of the specific crystalline compound, it would be obvious to recognize that specific ion-energy is to be applied thereon depending on the specific material composition and deposition techniques.

9. Claim 32 is rejected under 35 U.S.C. 103(a) as being unpatentable over Ovshinsky et al 5411592 in view of Bates et al 5338625 as applied to claim 11 above, and further in view of Matsui et al 5558953.

Ovshinsky et al and Bates et al are applied, argued and incorporated herein for the reasons above. In addition, the preceding references do not disclose the specific material LiCoO₂.

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Matsui et al disclose that as a positive electrode (cathode) of the lithium battery, it is

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preferable to use an active material such as LiCoO₂ (col 4, lines 48-57).

In view of the above, it would have been obvious to one skilled in the art at the time the

invention was made to use the specific material LiCoO₂ to form the second layer of the layer-

deposited battery of Ovshinsky et al and Bates et al as Matsui et al disclose that it is preferable to

use LiCoO₂ as an active material because such a compound is capable of imparting a discharge

voltage of 4V level.

Conclusion

Any inquiry concerning this communication or earlier communications from the

examiner should be directed to Raymond Alejandro whose telephone number is (703) 306-3326.

The examiner can normally be reached on Monday-Thursday (8:30 am - 7:00 pm).

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Patrick J. Ryan can be reached on (703) 308-2383. The fax phone numbers for the

organization where this application or proceeding is assigned are (703) 872-9310 for regular

communications and (703) 872-9311 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding

should be directed to the receptionist whose telephone number is (703) 308-0661.

Raymond Alejandro

Examiner

Art Unit 1745

Supervisory Patent Examiner

Technology Center 1700